

# A New Colorimetric Test for FeNa-EDTA in Flour



**FFI Workshop Arusha, Tanzania  
November 20, 2008**

**Dr. Carel Wreesmann**

**Akzo Nobel Functional Chemicals  
carel.wreesmann@akzonobel.com  
+31 6 22 49 69 20**

# Outline



AkzoNobel

Recommended iron fortificants for wheat flour

Methods to determine Fe contents

intrinsic Fe vs. fortification Fe

Phenanthroline (orange color) method (NEW)

FeNa-EDTA (Ferrazone®)

quality assurance

additional information

# Akzo Nobel: Key Figures



Net sales € 14 bln / year  
Number of employees approx. 60,000

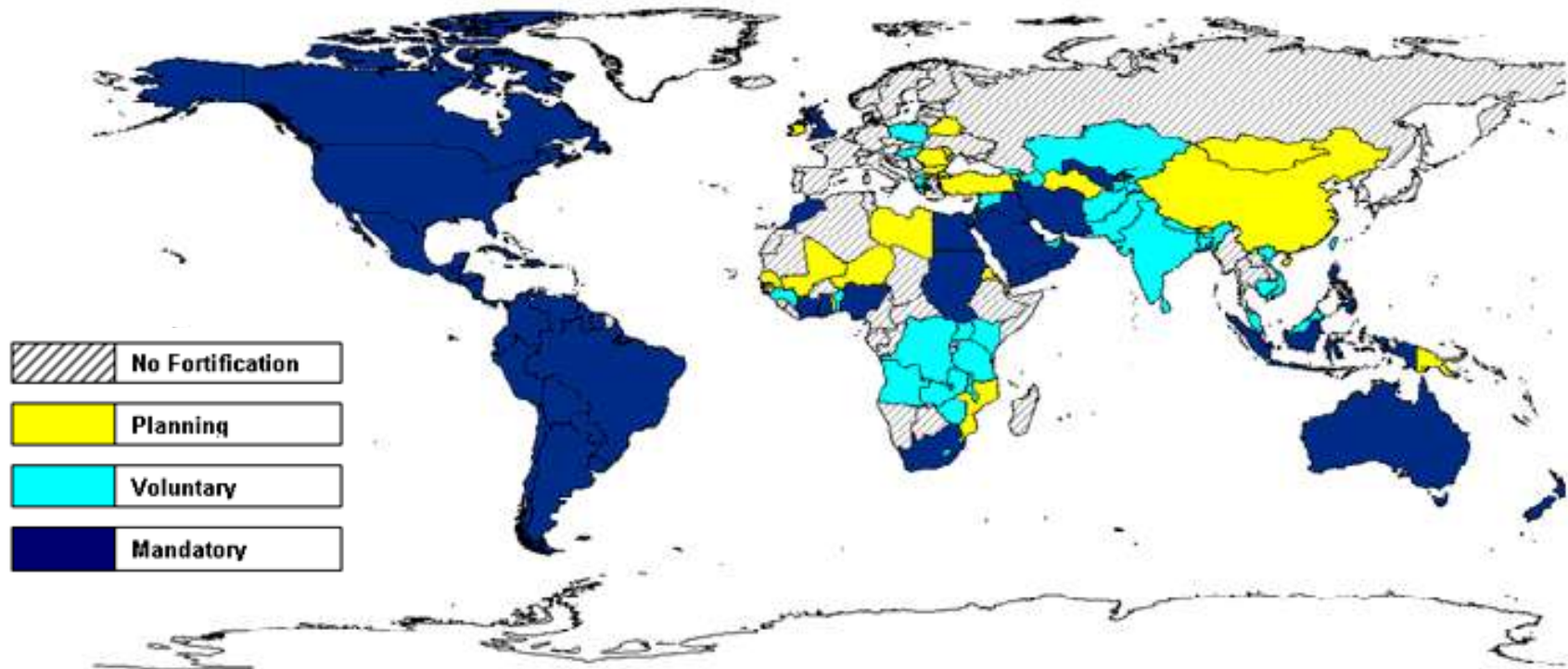
Decorative Paints  
Specialty Chemicals  
Performance Coatings



Ferrazone® → Specialty Chemicals  
Business Unit Functional Chemicals

# National Wheat Flour Fortification Programs

Fortification Status - September 2008



...about 2/3 of the countries in the world run wheat flour fortification programs...

Source: Flour Fortification Initiative (FFI) at

<http://www.sph.emory.edu/wheatflour/globalmap.php>

# Recommended Fe Fortification Levels



Second Technical Workshop on Flour Fortification  
US CDC / FFI, Atlanta – USA, April 2008

<http://www.sph.emory.edu/wheatflour/atlanta08/summary.html>

Type of flour	Fortificant	Average daily consumption in g/d		
		> 300	150 – 300	< 150
LOW-extraction	<b>FeNa-EDTA</b>	<b>15</b>	<b>20</b>	<b>40</b>
	Ferrous sulfate or Ferrous fumarate	20	30	60
	Electrolytic iron	40	60	not recommended
HIGH-extraction	<b>FeNa-EDTA</b>	<b>15</b>	<b>20</b>	<b>40</b>

# Intrinsic Fe Content in Flours

## High-extraction wheat flour

average: ~ 30 ppm

range: 10 – 100 ppm (?)

## Low-extraction wheat flour

average: ~ 10 ppm

range: 5 – 50 ppm (?)

Fe content may vary considerably from batch to batch  
and even within batches!



# Total Fe Content



## Determination

dissolution of the flour in  $\text{HNO}_3/\text{H}_2\text{O}_2$

ultrahigh temperature in microwave

all insoluble parts → fully soluble → clear solution

injection into the flame of AAX / ICP

Total [Fe] can vary from 20 – 160 ppm

[Fe] from fortification: 15 – 60 ppm

depending on iron compound and consumption level

[Fe] of intrinsic origin: 5 – 100 ppm

# Intrinsic vs. Fortification Fe?

How to differentiate intrinsic vs. fortification Fe?  
both types fully dissolve

Suppose: AAS/ICP → 40 ppm Fe  
10 ppm from fortification, 30 ppm intrinsic?  
or the other way round?  
or 20 ppm + 20 ppm?

Type of fortificant (bio-availability) is critically important

Total Fe content (AAS/ICP) is not suitable



# Other Methods for Flour

## Red Spot Test

directly on the flour as such  
only for elemental iron types

## Extraction of flour + colorimetric determination

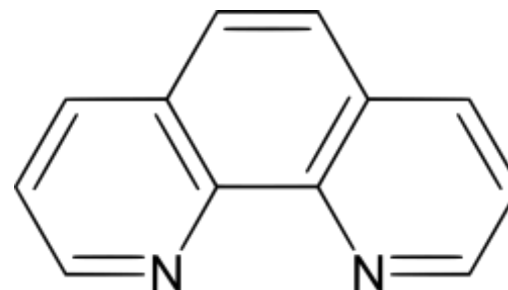
flour is extracted with water/methanol → suspension  
flour is separated from water layer  
coloring agents are added to filtrate solution

## Phenanthroline (orange color) method (NEW)

# Determination of [Fe] in Water

## Phenanthroline (Phen)

also: ortho-phenanthroline



Gives a highly intense, orange color with ferrous ions



Ferric ions should be reduced first (e.g. by ascorbic acid)



Analytical method already known from 1930's

Widely-used for [Fe] determination in water

# Semi-Quantitative Procedure (DEMO)

Mix 10 g of flour with 30 mL of water/methanol (80/20)  
shake vigorously during 1 minute

Separate water from flour over paper filter (~ 15 minutes)

Transfer 15 mL to 50-mL volumetric flask

Add the following reagents

5 mL vitamin C solution (1 g/L)

5 mL phenanthroline solution (1 g/L)

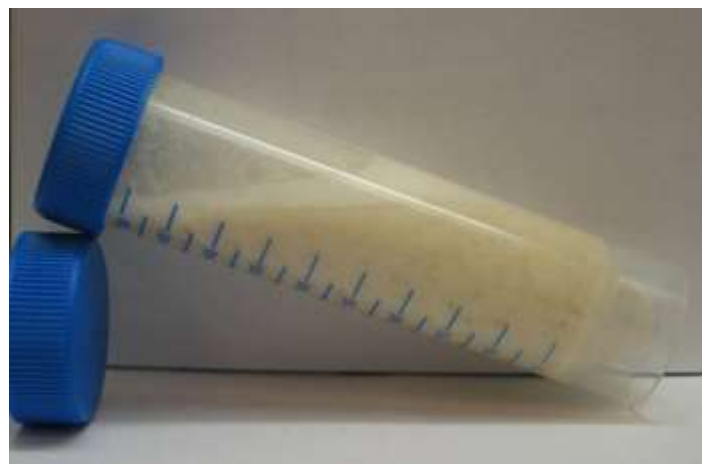
Fill up to 50-mL mark with water

Wait for 60 minutes

Assess color intensity by visual inspection with calibration samples

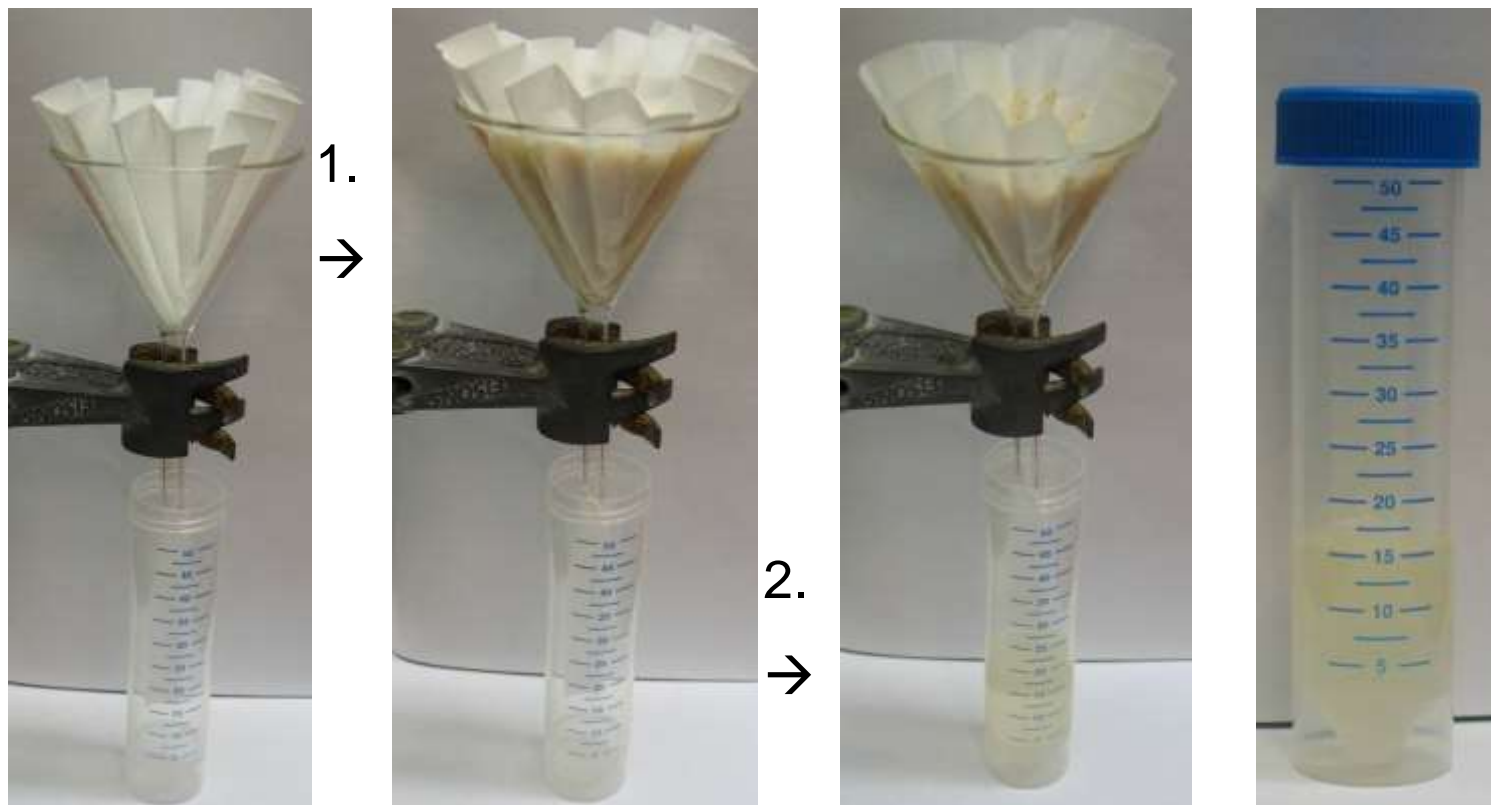
## Mix with Water/Methanol

Transfer 10 g of fortified flour into a 50-mL test tube  
Add 30 mL water/methanol (80/20) and mix thoroughly to suspend all flour  
Shake resulting suspension vigorously for 1 minute



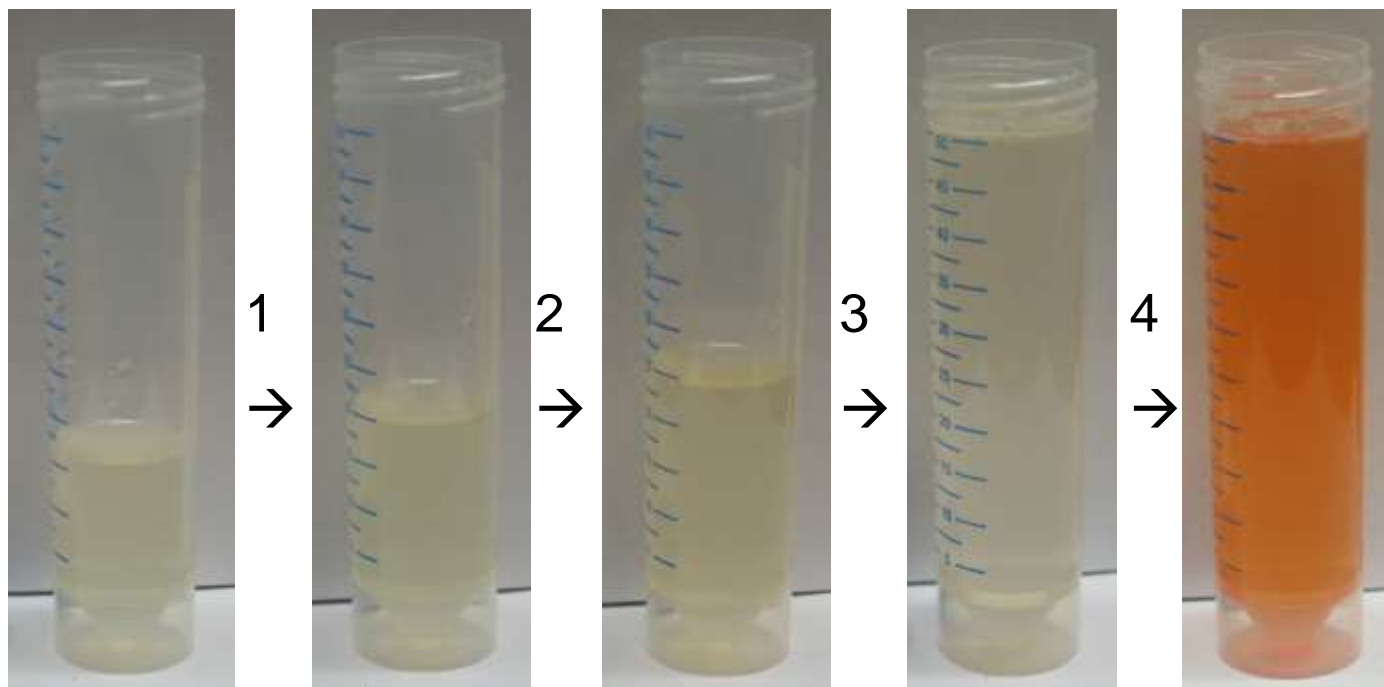
# Filter Solution

1. Separate water from flour through a folded paper filter
2. Collect 15 mL filtrate in a 50-mL test tube

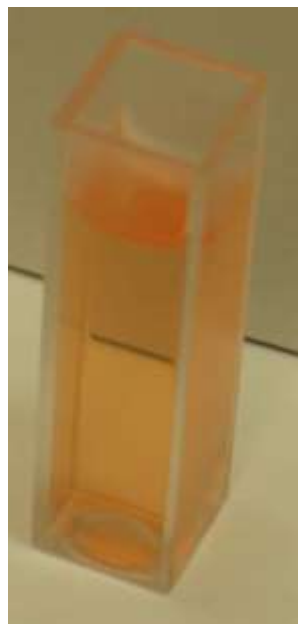


# Add Reagents

1. 5.0 ml of vitamin C solution (1 g/L)
2. 5.0 ml of phenanthroline solution (1 g/L)
3. Fill up to 50 mL and mix
4. After 1 hour the color reaction has completed

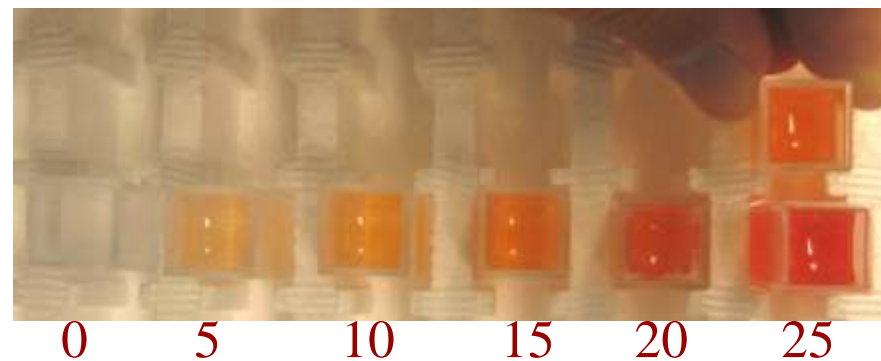
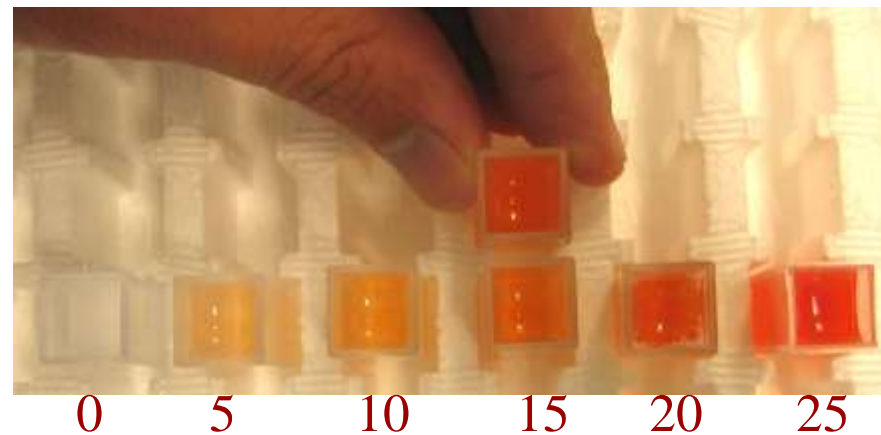
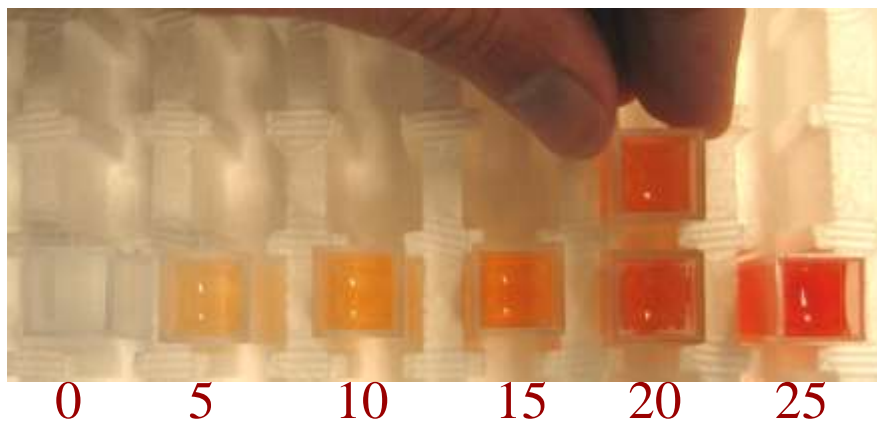


# Transfer to Cuvette





# and Compare Visually



# Other Iron Fortificants

## In **HIGH**-extraction flour

Fe as Ferrazone: can be determined reliably (2 – 30 ppm)

Fe as *dried*  $\text{FeSO}_4$ : is NOT detected (bound to phytate?)

even heptahydrate  $\text{FeSO}_4$  is hardly detected

Fe as ferrous fumarate: is not detected either

Fe as electrolytic iron: does not dissolve in water

***Conclusion: phenanthroline method is suitable for discrimination Ferrazone vs. other (recommended) iron fortificants in high extraction wheat flour***

# Ongoing Research



Demo = *semi*-quantitative method

Also quantitative method available

filtration → centrifuge

visual comparison → spectrophotometer

vitamin C → hydroxylamine

HIGH-extraction flours: methods work OK

other iron fortificants are not detected (phytate?)

also true for LOW-extraction flours (low in phytate)?

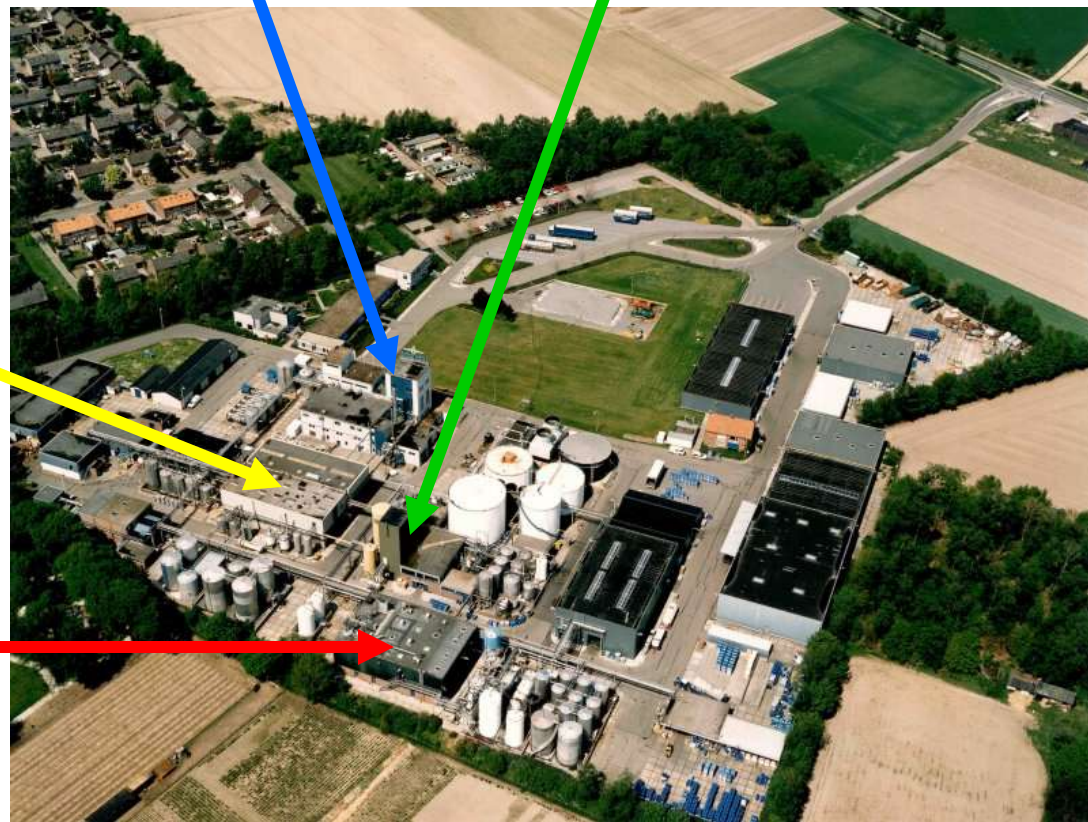
# Manufacturing & Quality Assurance

$\text{Na}_2\text{H}_2\text{-EDTA}$

$\text{CaNa}_2\text{-EDTA}$

$\text{Na}_4\text{-EDTA}$

$\text{FeNa-EDTA}$



Production site in Herkenbosch, the Netherlands



# Cross Contamination

Major risk in food, pharma ingredients  
inevitable with >1 product on same line  
dedicated line >> multipurpose unit !  
rigorous cleaning !

Ferrazone: DTPA on same line  
toxicology DTPA  $\approx$  EDTA  
low risk → no “Critical Control Point” (CCP)

## Hazard Analysis Critical Control Points

how to prevent illness and worse from food?

Codex Alimentarius = WHO/FAO = worldwide

## Food Safety System

guidelines to check and recheck production

audit by independent bureau => certificate

CCP: parts in the process with enhanced health risk

e.g. micro-organisms in process water => extra controls

# Certificates

HACCP since February 2004

Also for Dissolvine E-CA-10  
and Dissolvine NA-2-P

Halal correct (below)





# Why Food Grade?

Cross contamination (see above)

FeNa-EDTA

also used, and at a large scale, in agriculture!

Plants do not care about sand, dust, glass splinters, ...  
humans do!

Starting materials

should be 100% free of toxic contaminants

e.g. hydrochloric acid (HCl)

may contain traces of chlorinated phenols due to recycling  
only “virgin” HCl should be used, made from  $H_2$  and  $Cl_2$ .

# Unsuitable FeNa-EDTA

FeNa-EDTA + water => clear brown solution

no insoluble matter = sand, dust, glass splinters, ...?

Filter test: paper should be entirely **white** after filtration

Brown precipitate and/or unidentifiable particles on filter

FeNa-EDTA for agricultural use only



# Regulatory Status FeNa-EDTA



JECFA (2007)

*Sodium iron EDTA is suitable for use as a source of iron for food fortification to fulfill nutritional iron requirements ...*

US FDA

Gras Notices GRN 152 and 178

EU EFSA

Dossier submitted, final statement pending

India

Approved for atta flour and drinks by PFA

Approval also in Brazil, China, Mexico, Pakistan, Philippines, Vietnam, ...

No formal approval yet in African countries

# Current Applications

China: soy sauce, wheat flour  
voluntary program with strong government support  
locally sourced FeNa-EDTA

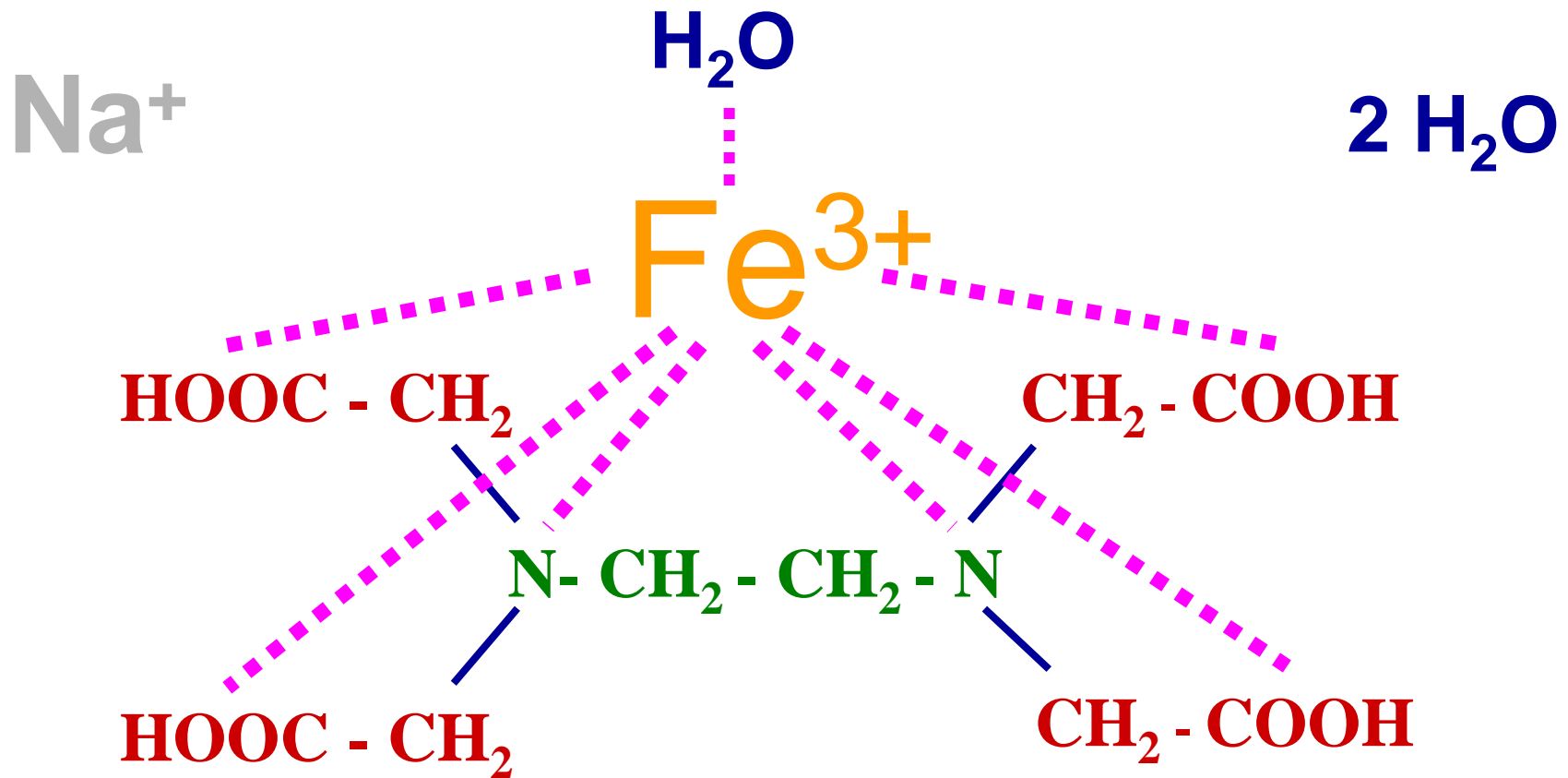
Vietnam: fish sauce (just launched)

Pakistan: atta flour (started in 2007)

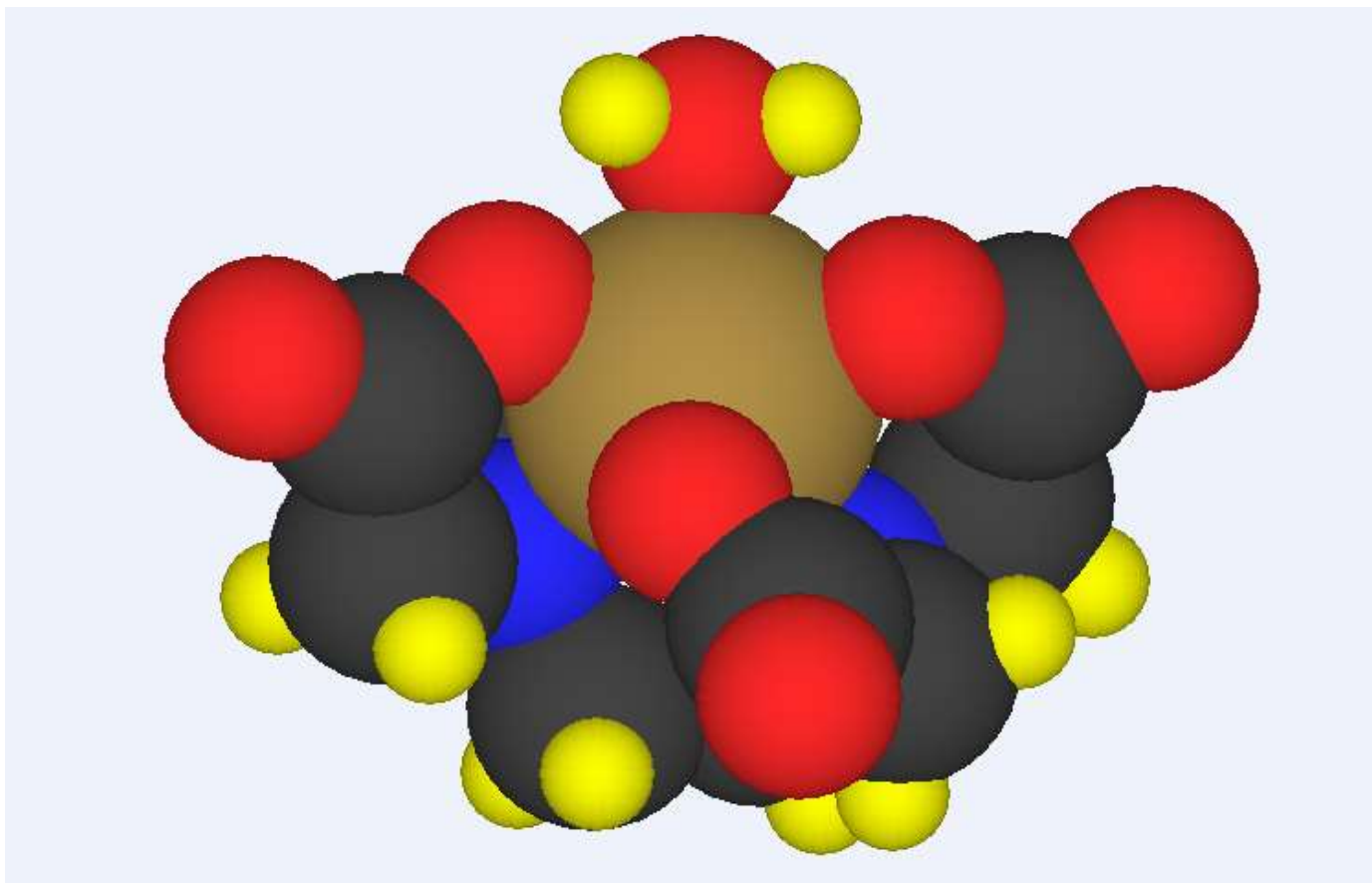
Brazil, Mexico, Philippines, ...:  
powdered beverage (Tang) of Kraft



# Ferric Sodium EDTA



# 3-Dimensional Model



Ferric-EDTA monohydrate complex-ion  
Wageningen University 2007



# Ferrazone® XF



Food-grade ferric sodium EDTA ex AkzoNobel

➔ Ferrazone®

Wheat flour: particle size < 150 micron  
special “extra fine” grade

➔ Ferrazone® XF

**Ferrazone® XF**  
**Sodium Iron (III) Ethylenediaminetetraacetate**

Ferrazone® XF is a stable, water soluble iron compound that meets JECFA specifications for food fortification. Ferrazone® XF is Generally Recognized As Safe (GRAS) by the US FDA. Ferrazone® XF manufacturing is certified to be in accordance with the HACCP requirements.

Checkpoint	Specification	Units	Method
Appearance	Light yellow coloured powder		visual
Iron content	12.5-13.5	%	JECFA
EDTA content	65.5-70.5	%	JECFA
Identification	Passes tests		JECFA
pH of a 1% w/v aqueous solution	3.5-5.5		JECFA
Water-insoluble matter	0.1 max	%	JECFA
Nitritriacetic acid	< 0.1	%	JECFA
Arsenic	1 max.	mg/kg	JECFA
Lead	1 max.	mg/kg	JECFA

Net Weight: 25 kg

LOT CODE

Produced in the Netherlands  
Akzo Nobel Functional Chemicals b.v.  
According to ISO 9002 & ISO 14001  
Keep out of direct sunlight  
Emergency Telephone  
Tel: + 31 570 679211 Fax: + 31 570 679801



# A New Colorimetric Test for FeNa-EDTA in Flour



***Thank You for Your Attention***

**Dr. Carel Wreesmann**

Akzo Nobel Functional Chemicals  
carel.wreesmann@akzonobel-chemicals.com  
+31 6 22 49 69 20